

# National Bureau of Standards

## Certificate of Analyses

OF

### STANDARD SAMPLE 9D

### BESSEMER STEEL, 0.2% CARBON

ANALYST*	C	Mn		P		S		Si	COPPER H <sub>2</sub> S-CuS-CuO	NICKEL Weighed as nickel dimethylglyoxime	CHROMIUM FeSO <sub>4</sub> -KMnO <sub>4</sub> titration	VANADIUM	MOLYBDENUM Colorimetric	NITROGEN
	Direct combustion	Bismuthate (FeSO <sub>4</sub> -KMnO <sub>4</sub> )	Persulfate-Arsenite	Gravimetric (weighed as MgP <sub>2</sub> O <sub>7</sub> after removal of arsenic)	Alkali-Molybdate <sup>a</sup>	Gravimetric (direct oxidation and final precipitation in reduced solution)	Evolution with HCl (1-1) ZnS-Iodine (theoretical sulfur titre) <sup>b</sup>	Sulfuric acid dehydration						
1	0. 200	0. 618	0. 616	0. 095	0. 097	0. 035	0. 036	0. 033	0. 008	0. 004	0. 004	0. 006	0. 001	0. 018
2	. 198		. 62	h. 100	. 099	. 039	. 038	i. 029	. 007	. 004	. 003	i. 005	. 003	
3	. 200		k. 623		k. 097		. 039	. 030	l. 011	m. 003	m. 005	m. 006	m. 001	p. 015
4	. 205		k. 621		k. 098	. 036	. 034	od. 033	m. 009					
5	. 206	. 624	. 62	. 096	. 097	. 035	p. 035	d. 036	. 012	. 004	e. 005	f. 005	. 001	q. 017
	. 204	. 618	. 619	. 095	. 096	. 036	. 036	. 035	. 010	m. 004	m. 004		. 001	q. 018
	. 207		k. 620		k. 099		k. 037	i. 030	r. 008	. 005	s. 004	i. 004	. 002	
8	. 207		. 620	h. 095	. 096	. 036	. 036	od. 032	. 011	m. 005	. 007	k. 01	. 003	p. 017
9	. 205		k. 620	. 096	k. 097	. 038	k. 038	id. 033	. 005	. 003	. 004	. 004	. 001	t. 016
10	. 208		. 62		. 098	. 035	. 037	. 031	m. 007	m. 004	. 004	i. 004		
Averages	0. 204	0. 620	0. 620	0. 096	0. 097	0. 036	0. 037	0. 032	0. 009	0. 004	0. 004	0. 005	0. 002	0. 017
General average	0. 204	0. 620		0. 096		0. 036		0. 032		0. 004		0. 005	0. 002	0. 017

<sup>a</sup> Precipitated at 40° C, washed with a 1-percent solution of KNO<sub>3</sub> and titrated with alkali standardized by the use of National Bureau of Standards acid potassium phthalate and the ratio 23NaOH:1 P.

<sup>b</sup> Value obtained by standardizing the titrating solution by means of sodium oxalate through KMnO<sub>4</sub> and Na<sub>2</sub>S<sub>2</sub>O<sub>8</sub>, and use of the ratio 21:1S.

<sup>c</sup> Colorimetric method. See J. Research NBS 26, 405 (1941) RP1386.

<sup>d</sup> Double dehydration.

<sup>e</sup> Persulfate oxidation and potentiometric titration with ferrous ammonium sulfate solution standardized

with recrystallized potassium dichromate.

<sup>f</sup> Nitric acid oxidation and potentiometric titration with ferrous ammonium sulfate solution standardized with recrystallized potassium dichromate.

<sup>g</sup> Determination made by M. Marie Cron, by the vacuum-fusion method. See NBS J. Research 7, 375 (1931) RP346.

<sup>h</sup> Weighed as ammonium phosphomolybdate.

<sup>i</sup> Perchloric acid dehydration.

<sup>j</sup> Ferrous sulfate-persulfate-KMnO<sub>4</sub> titration.

<sup>k</sup> Titrating solution standardized by use of a standard steel.

<sup>l</sup> Finished by electrolysis.

<sup>m</sup> Colorimetric method.

<sup>n</sup> Allen method.

<sup>o</sup> Nitric-sulfuric acid dehydration.

<sup>p</sup> Titrating solution standardized by use of an empirical factor.

<sup>q</sup> Determined colorimetrically following semimicro-distillation.

<sup>r</sup> KI-Na<sub>2</sub>S<sub>2</sub>O<sub>8</sub> titration.

<sup>s</sup> Perchloric acid oxidation.

<sup>t</sup> Solution in sulfuric-phosphoric acid mixture. Distillation and titration.

#### \*LIST OF ANALYSTS

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The steel for the preparation of this standard was furnished by the Jones & Laughlin Steel Corporation.

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