Certificate of Analysis

Standard Reference Material 340

Ferroniobium

ANALYST	Nb ^a	Ta ^b	Ti ^c	С	Mn	P	Si	Sn
1	57.51	3.67	0.87					
2	57.55	3.78	.87	0.059		0.037	4.42	
3	57.56	3.66	.90	.055	1.72	.035	4.36	0.063
4	57.50	3.76	.90	.063	1.69	.030		
5	57.53	3.79	.90	.066	1.79		4.45	.066
6	57.42	3.78	.89	.057	1.71	.039	4.37	.057
7	57.57	3.70	.87		1.64	.035	4.36	.065
8	57.47	3.76	.89					
9	57.47	3.66	.84	.066	1.66	.040		
Average	57.51	3.73	0.89	0.061	1.70	0.036	4.39	0.063

^a Sample dissolved in HCl-HF mixture and transferred to ion-exchange column. After removal, by elution, of titanium, iron, etc., the niobium is removed from the column by eluting with NH₄Cl-HF solution, treated with boric acid, precipitated with cupferron, ignited and weighed as Nb₂O₅.

solution adjusted to pH 5-6, treated with boric acid, precipitated with cupferron, ignited and weighed as Ta₂O₅.

List of Analysts

- 1. L. A. Machlan, Analytical Chemistry Division, Institute for Materials Research, National Bureau of Standards.
- 2. E. Jacobson, U. S. Army Materials and Mechanics Research Center, Watertown, Massachusetts.
- 3. S. Kallman, Ledoux and Company, Teaneck, New Jersey.
- 4. A. Thomas, Armco Steel Corporation, Middletown, Ohio.
- 5. J. B. Armstrong, Bethlehem Steel Corporation, Sparrows Point Plant, Sparrows Point, Maryland.
- 6. G. Porter, Union Carbide Metals Company, Niagara Falls, New York.
- 7. L. Risi, Shieldalloy Corporation, Newfield, New Jersey.
- 8. E. A. Lucas, Molybdenum Corporation, Washington, Pennsylvania.
- 9. S. Brown, Tennessee Products, Chattanooga, Tennessee.

The material for this standard was furnished by the Electro-metallurgical Corporation, Division of Union Carbide Corporation, Niagara Falls, New York.

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^bSample solution and removal of titanium as in (a), tantalum removed from the column by eluting with NH₄Cl-NH₄F

^cSample solution as in (a), titanium eluted with NH₄Cl-HCl-HF solution, treated with boric acid, precipitated with cupferron, ignited, fused with KHSO₄ and leached in dilute $\rm H_2SO_4$. Titanium determined spectrophotometrically with $\rm H_2O_2$.