

Bureau of Standards

Certificate of Analyses

OF

STANDARD SAMPLE No. 21b
ACID OPEN-HEARTH STEEL, 0.6% CARBON

ANALYST.*	C	Mn		P		S		Si	COPPER. H ₂ S-CuS-CuO.	NICKEL. Weighed as nickel dimethylglyoxime.	CHROMIUM. FeSO ₄ -KMnO ₄ titration.	VANADIUM.	MOLYBDENUM.	ARSENIC.
	CARBON. Direct combustion.	MANGANESE. 1. Bismuthate (FeSO ₄ -KMnO ₄), 2. Other methods.	PHOSPHORUS. 1. Alkali-Molybdate. ^a 2. Gravimetric (Weighed as MgP ₂ O ₇ after removal of arsenic).	1. SULPHUR. Gravimetric (Direct oxidation and final precipitation in reduced solution), 2. SULPHUR. Evolution with HCl (1:1) ZnS-Iodine (theoretical sulphur titre ^b).	SILICON. Sulphuric acid dehydration.									
1.....	0.608	0.560		0.065	0.063	0.037	0.037	0.163	0.031	0.060	0.024	0.004	0.005	0.01
2.....	.606	.564		.064	.063	.037	.036	.161	.031	.060	.025	.005	.007	.008
3.....	.602	.572	0.565 ^c	.062		.039	.038	{.162 ^d .162 }	.037 ^e	.06		.007		
4.....	.604	.569	.57 ^c	.065	.064	.035	.035	.159	.028	.063			.005	
5.....	.609	.558		.065		{.034 ^f .034 }	.034	.157						
6.....	.591		.566 ^g	.063		.035	.035	{.164 ^h .162 ⁱ }	.026	.064 ^j				
7.....	.616		.554 ^k	.065		.035		.164	.033 ^l	.065 ^l	.015			
8.....	.600		.563 ^c	.065		.036	.035	.162			.017			
Averages	.605	.565	.564	.064	.063	.036	.036	.162	.031	.062	.020	.005	.006	.009
General Averages...	.605	.564		.064		.036		.162	.031	.062	.020	.005	.006	.009

^a Precipitated at 40° C., washed with 1 per cent KNO₃ solution and titrated with alkali standardized by the use of B. S. benzoic acid and the 23:1 ratio.
^b Value obtained by standardization of titrating solution against sodium oxalate through KMnO₄ and Na₂S₂O₃.

^c Persulphate-arsenite.
^d Solution in HCl and dehydration with H₂SO₄.
^e Finished by electrolysis.
^f Precipitated in FeCl₃ solution.
^g Bismuthate-arsenite.
^h HCl dehydration.

ⁱ Solution in HNO₃ and H₂SO₄.
^j Weighed as NiO.
^k PbO₂-arsenite.
^l Finished by titration with KCN.

*LIST OF ANALYSTS

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8. R. B. Hooper, Dodge Brothers, Detroit, Mich.

This standard is not recommended for colorimetric carbon determinations, because of uncertainty as to the condition of the carbon.

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